Montiporic Acids A and B, Cytotoxic and Antimicrobial Polyacetylene Carboxylic Acids from Eggs of the Scleractinian Coral Montipora digitata¹

Nobuhiro Fusetani,*,[†] Takeo Toyoda,[†] Naoki Asai,[†] Shigeki Matsunaga,[†] and Tadashi Maruyama[‡]

Laboratory of Aquatic Natural Products Chemistry, Graduate School of Agricultural and Life Sciences, The University of Tokyo, Bunkyo-ku, Tokyo 113, Japan, and Marine Biotechnology Institute, Shimizu Laboratories, Sodeshi-cho, Shimizu-shi, Shizuoka 424, Japan

Received April 22, 1996[®]

Two new polyacetylene carboxylic acids, montiporic acids A (1) and B (2), have been isolated from the eggs of the scleractinian coral Montipora digitata and their structures elucidated on the basis of spectroscopic data. They exhibited antimicrobial activity against *Escherichia coli* and cytotoxicity against P-388 murine leukemia cells.

Montipora digitata Dana, 1845 (Scleractinia: Coelenterata) is a hermaphroditic coral; its colonies release bundles of tangled eggs and sperm, which are disentangled prior to fertilization. Polyacetylene alcohols are released from the eggs to attract sperm.² Because large numbers of eggs are released during the full moon in early summer, it is likely that the eggs are capable of chemical defense against predators,^{3–7} which prompted us to examine bioactive constituents of the released eggs of M. digitata. Indeed, the EtOH extract showed antibacterial and cytotoxic activities. Bioassay-guided isolation afforded two polyacetylene carboxylic acids. This paper describes the isolation and structure elucidation of these compounds.

The CH₂Cl₂-soluble portion of the aqueous EtOH extract of the frozen eggs (3.0 g) was partitioned between aqueous MeOH and *n*-hexane, followed by CH₂-Cl₂. The aqueous MeOH layer, which showed antibacterial activity against Escherichia coli, was purified by repeated chromatography on Si gel and ODS to yield montiporic acids A (1, 0.5 mg) and B (2, 0.9 mg).



Montiporic acid A (1) had a molecular formula of C₁₄H₂₀O₃ as established by HRFABMS. The IR spectrum exhibited bands of a disubstituted acetylene (2250 cm^{-1}) and a carboxyl (1730 cm^{-1}) group. The ¹H- and ¹³C-NMR spectra (Table 1) revealed a terminal methyl ($\delta_{\rm H}$ 0.86; $\delta_{\rm C}$ 13.9), two oxygenated methylenes ($\delta_{\rm H}$ 4.36, 4.22; $\delta_{\rm C}$ 59.1, 65.8), two acetylenic units ($\delta_{\rm C}$ 82.1, 72.9, 69.7, 65.8), and a carboxylic acid ($\delta_{\rm C}$ 178.8). ¹³C signals were reminiscent of a conjugated diyne system, which was supported by IR and UV data.8,9 HMBC cross peaks H1/C2, H1/C3, H1/C4, H6/C3, H6/C4, and H6/C5 revealed that the conjugated diyne was connected to C1

Table 1. ¹H- and ¹³C-NMR Data for Montiporic acids A (1) and B (2) in CDCl₃

	1		2	
position	$^{1}\mathrm{H}$	¹³ C	¹ H	¹³ C
1	4.36 (2H, s)	59.1 (t)	4.36 (2H, s)	59.1
2		69.7 (s)		69.7
3		72.9 (s)		73.0
4		64.2 (s)		64.2
5		82.1 (s)		82.3
6	2.26 (2H, t, J = 7.5 Hz)	19.0 (t)	2.22 (2H, t, J = 7.5 Hz)	19.2
7	1.52 (2H, quint., J = 7.5 Hz)	27.9 (t)	1.47 (2H, quint., J = 7.5 Hz)	28.0
8	1.34 (2H, qunit., $J = 7.5$ Hz)	28.4 (t)	1.35 (2H, quint., J = 7.5 Hz)	28.9
9	1.24 (m)	28.4 (t)	1.25 (m)	28.9
10	1.24 (m)	31.6 (t)	1.25 (m)	28.9
11	1.25 (m)	22.6 (t)	1.35 (m)	28.9
12	0.86 (3H, t, J = 7.0 Hz)	13.9 (q)	1.97 (2H, tdt, J = 8.0, 6.7,	33.7
13			1.5 Hz) 5.79 (1H, ddt, J = 17.0, 10.2, 6.7 Hz)	139.1
14a			4.98 (1H, ddt, J = 17.0, 2.1, ddt)	114.2
14b			1.5 Hz) 4.92 (1H, ddt, J = 10.2, 2.1,	
11	1 99 (9 Н с)	65.8 (+)	1.4 ПZ) 1.99 (9Ц с)	65.9
2'	4.22 (211, 5)	178.8 (s)	4.22 (211, 5)	172.9

and C6 methylene carbons. It must be noted that the fortuitous observation of four-bond couplings through the acetylenic bond allowed the unambiguous assignment of this system. HMBC cross peaks (H1/C1', H1'/ C1) and a NOESY correlation (H1/H1') displayed further connection of C1 to C1' through an ether oxygen. C1' was a part of substituted acetic acid unit, because it was correlated with the C2' carboxylate carbon in the HMBC spectrum. The remaining portion was an unbranched C_6 -alkane linked to C6, which was supported by the 2D-NMR data. Thus, montiporic acid A is 2-O-(2,4-dodecadiynyl)ethanolic acid. The fragmentation pattern observed in the negative FABMS/MS spectrum was consistent with this structure (Figure 1).

Montiporic acid B (2) had a molecular formula of C₁₆H₂₂O₃, which was determined by HRFABMS. Its ¹H-NMR spectrum was superimposable on that of 1 (Table 1), except for the presence of a monosubstituted olefin ($\delta_{\rm H}$ 5.79, 4.98, and 4.92) instead of a terminal methyl group. The presence of the conjugated diyne and

^{*} To whom correspondence should be addressed. Phone: $+81\mathcase -81\mathcase -8$ ecc.u-tokyo.ac.jp.

Laboratory of Aquatic Natural Products Chemistry.

 [‡] Marine Biotechnology Institute.
 [®] Abstract published in Advance ACS Abstracts, August 1, 1996.

Figure 1. Prominent fragment ions in the FAB-MS/MS spectrum of 1.

glycolate ether was straightforward from the HMBC spectrum. Therefore, montiporic acid B is 2-O-(13tetradecane-2,4-diynyl)ethanolic acid.

Montiporic acids A and B were not only antibacterial against Escherichia coli, but also cytotoxic against P-388 murine leukemia cells with IC₅₀ values of 5.0 and 12.0 μ g/mL, respectively. These compounds are closely related to the methyl esters reported from an Okinawan Montipora sp.;8 their spectral data are consistent with those of our compounds.

Experimental Section

General Experimental Procedures. NMR spectra were recorded either on a Bruker AM 600 or a Bruker AC 300 NMR spectrometer. Chemical shifts were referenced to solvent peaks: $\delta_{\rm H}$ 7.24 and $\delta_{\rm C}$ 77.0 for CDCl₃. MS were measured with a JEOL SX-102 mass spectrometer. Triethanolamine was used as a matrix in the FABMS, HRFABMS, and FABMS/MS.

Animal Material. Eggs of *M. digitata* were collected at Orpheus Island, Australia, as described previously.¹⁰

Extraction and Isolation. The frozen eggs (3.0 g) thus obtained were homogenized and extracted three times with EtOH/H₂O (7:3). The combined extracts were concentrated and partitioned between CH₂Cl₂ and H₂O. The organic layer was dissolved in MeOH/H₂O (9:1) and extracted with *n*-hexane, and the aqueous MeOH phase was further extracted with CH₂Cl₂. The aqueous MeOH layer was fractionated by ODS flash chromatography (H₂O/MeOH/CHCl₃, stepwise gradient). The fraction eluted with 100% MeOH was chromatographed on a Si gel column with a CHCl₃/MeOH system, followed by preparative TLC on Si gel with CHCl₃/ MeOH/H₂O (70:30:5). The active band was finally purified by repeated HPLC on an ODS column (MeOH/ H₂O, 9:1, MeCN/H₂O, 7:3) to afford montiporic acids A (1, 0.5 mg) and B (2, 0.9 mg).

Montiporic acid A (1): colorless oil; UV (MeOH) λ max 255 (ϵ 1059), 273 (799); IR (film) ν_{max} 2925, 2850, 2250, 1730 cm⁻¹; HRFABMS (negative) m/z found 235.1332 (M – H)⁻, calcd for $C_{14}H_{19}O_3$ 235.1334; ¹H and ¹³C NMR data, see Table 1.

Montiporic acid B (2): colorless oil; UV (MeOH) λ max 256 (ϵ 861), 273 (668); IR (film) ν max 3070, 2925, 2850, 2250, 1730, 1640 cm⁻¹; HRFABMS (negative) m/zfound 261.1500 (M – H)⁻, calcd for $C_{16}H_{21}O_3$ 261.1491; ¹H and ¹³C NMR data, see Table 1.

Acknowledgment. We are grateful to Professor Paul J. Scheuer, University of Hawaii, for reading this manuscript. We are indebted to Dr. S. Fukuzawa for cytotoxicity tests and to Dr. N. Sata for assistance in measuring HRFABMS and FABMS/MS.

References and Notes

- Bioactive Marine Metabolites. 77. Part 76: Ryu G.; Matsunaga S.; Fusetani N. *J. Nat. Prod.* **1996**, *59*, 515–517.
 Coll, J. C.; Bowden, B. F.; Meehan, G. V.; König, G. M.; Carroll, A. R.; Tapiolas, D. M.; Alino, P. M.; Heaton, A.; Nys, De R.; Leone, P. A.; Maida, M.; Aceret, T. L.; Willis, R. H.; Babcock, R. C.; Willis, B. L.; Florian, Z.; Clayton, M. N.; Miller, R. L. *Mar. Biol. (Berlin)* **1994**, *118*, 177–182.
- (3) Faulkner, D. J. Nat. Prod. Rep. 1995, 12, 233-269 and references cited therein.
- (4) Coll, J. C. Chem. Rev. 1992, 92, 613-631.
- Sammarco, P. W.; Coll, J. C. In *Bioorganic Marine Chemistry*, Scheuer, P. J., Ed.; Springer-Verlag: New York, 1988; Vol. 2, (5)
- pp 87–116. (6) Fusetani, N.; Asano, M.; Matsunaga, S.; Hashimoto, K. *Comp.* Biochem. Physiol. 1986, 85B, 845-846.
- (7) Sakai, R.; Higa, T. Chem. Lett. 1987, 127-128.
- (8) Higa, T.; Tanaka, J.; Kohagura, T.; Wauke, T. Chem. Lett. 1990, 145-148.
- (9) Scott, A. I. Interpretation of the Ultraviolet Spectra of Natural Products; Pergamon Press: London, 1964; p 283.
- (10) Arai, T.; Kato, M.; Heyward, A.; Ikeda, Y.; Iizuka, T.; Maruyama, T. Coral Reefs 1993, 12, 71-75.

NP9604036